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**Open**

Tick (✓) true or false for each statement.

<b>1. Neutralization reaction is accompanied by evolution of heat.</b> 1. True 2. False	<b>2. Neutralization reaction is a reaction between an acid and a base to form salt and water.</b> 1. True 2. False
<b>3. Turmeric is a natural indicator.</b> 1. True 2. False	<b>4. Turmeric acquires orange colour in alkaline solution.</b> 1. True 2. False
<b>5. Onions cannot be used as indicators.</b> 1. True 2. False	<b>6. Litmus is a natural dye and is extracted from papyrus.</b> 1. True 2. False
<b>7. Limewater turns milky when we breathe through it.</b> 1. True 2. False	<b>8. Red litmus turns blue in a solution containing vitamin C.</b> 1. True 2. False
<b>9. Vinegar is acidic in nature.</b> 1. True 2. False	<b>10. Alkali is a base which is not soluble in water.</b> 1. True 2. False

LC6127

**Reactions of Acids and Bases**

**Reactions of Acids**

- With bases - give salt and water



- With active metal - to liberate hydrogen gas



- With carbonates and bicarbonates - to liberate carbon dioxide gas



**Reactions of Bases**

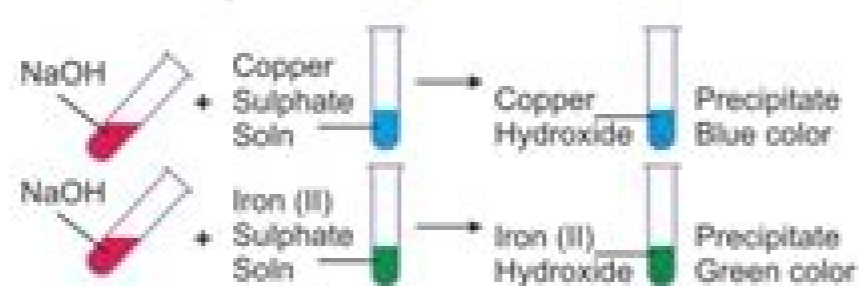
- With acids – give salt and water



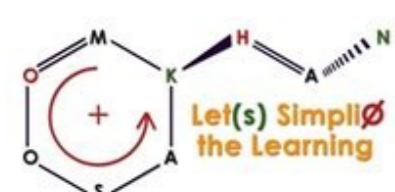
- With ammonia – give ammonia gas



- With soluble metallic salts giving metallic hydroxide precipitate



O Level Chemistry  
Topic: Acid-Base and Salts Class 1

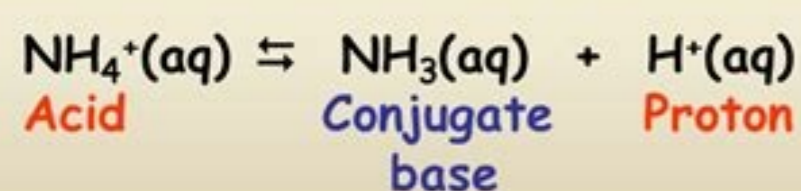


O/A level Chemistry  
With  
Moosa BH Khan

**Acid-Base Properties of Salts**

Type of Salt	Examples	Comment	pH of solution
Cation is the conjugate acid of a weak base, anion is from a strong acid	NH <sub>4</sub> Cl, NH <sub>4</sub> NO <sub>3</sub>	Cation is acidic, Anion is neutral	Acidic

The acidic cation can act as a proton donor:







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